

## Safety Data Sheet

### 1. IDENTIFICATION

<b>Product Identifier:</b>	<b>Sulfuric Acid, 0.94-6N, 5-12% v/v, 25% w/w</b>
<b>Product Code(s):</b>	S1060, S1061, S1062, S1064, S1114, S1115, S1116, S1118, S1173, S1187, CF1110
<b>Synonyms:</b>	Hydrogen Sulfate; Oil of Vitriol; Babcock Acid
<b>Recommended Use:</b>	For manufacturing, industrial, and laboratory use only. Use for neutralization of basic systems, as a catalyst, as a solvent, or as a laboratory reagent.
<b>Uses Advised Against:</b>	Not for food, drug, or household use.
<b>Supplier:</b>	Rocky Mountain Reagents, Inc. 4621 Technology Drive, Golden, CO 80403 Phone: (303) 762-0800 Fax: (303) 762-1240
<b>Emergency Phone Number:</b>	(800) 255-3924 (CHEM-TEL)

### 2. HAZARDS IDENTIFICATION

<b>Hazard Classifications:</b>	Acute Toxicity – Inhalation: Category 4
	Skin Corrosion/Irritation: Category 1A
	Eye Damage/Irritation: Category 1

**Signal Word:** DANGER

**Hazard Statements:** Harmful if inhaled.  
Causes severe skin burns and serious eye damage.

**Pictograms:**



**Precautionary Statements:**

**Prevention:** Do not breathe fumes, mists, vapors, or spray.  
Use only outdoors or in a well-ventilated area.  
Wash thoroughly after handling.  
Wear protective gloves, protective clothing, eye protection, and face protection.

**Response:** Immediately call a poison center or doctor.  
If inhaled: Remove person to fresh air and keep comfortable for breathing.

If swallowed: Rinse mouth. Do NOT induce vomiting.  
If on skin (or hair): Take off immediately all contaminated clothing. Rinse skin with water.  
Wash contaminated clothing before reuse.  
If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

**Storage:** Store locked up.

**Disposal:** Dispose of contents and container in accordance with local, regional, national, and international regulations.

**Hazards Not Otherwise Classified:** This product is hazardous to the environment. Avoid release to groundwater or aquatic environments.  
Inhalation of inorganic mists containing sulfuric acid may cause cancer.

**Toxicity Statement:** Not applicable.

### 3. COMPOSITION AND INFORMATION ON INGREDIENTS

Component	Common Name / Synonyms	CAS#	Chemical Formula	% by Weight
Sulfuric Acid	Hydrogen Sulfate, Oil of Vitriol	7664-93-9	H <sub>2</sub> SO <sub>4</sub>	5 – 24
Water	Water	7732-18-5	H <sub>2</sub> O	76 – 95

**Trade Secret Statement:** Not applicable.

### 4. FIRST AID MEASURES

#### First Aid Procedures:

**Inhalation:** Move to fresh air. If breathing is difficult, give oxygen. If not breathing, give artificial respiration. WARNING! It may be hazardous to the person providing aid to give mouth-to-mouth resuscitation when the inhaled or ingested material is toxic, infectious, or corrosive. Do not use mouth-to-mouth resuscitation if victim ingested or inhaled the substance; induce artificial respiration with the aid of a pocket mask equipped with a one-way valve or other proper respiratory medical device. Call a physician immediately.

**Ingestion:** Do not induce vomiting unless directed to do so by medical personnel. If vomiting occurs, keep head low so that vomit does not enter lungs. Never give anything by mouth to an unconscious person. Immediate medical attention is required. Call a physician or poison control center immediately.

**Skin Contact:** Wash skin with soap and plenty of water for at least 15 minutes. Remove contaminated clothing and shoes. Wash clothing before reuse. Call a physician immediately.

**Eye Contact:** Check for and remove contact lenses, if present and easy to do. Immediately flush eyes with gentle but large stream of water for at least 15 minutes, lifting lower and upper eyelids occasionally. Immediate medical attention is required. Call a physician immediately.

**General Advice:** Poison information centers in each state can provide additional assistance for scheduled poisons. Ensure that those providing first aid and medical personnel are aware of the material(s) involved and take precautions to protect themselves.

**Symptoms and Effects:** Severe skin and eye irritation or burns, irritation of respiratory system, burning sensation of the respiratory tract, coughing, hoarseness, choking sensation, dyspnea (shortness of breath and difficulty breathing), shallow respiration, salivation, burning of mouth, throat, and

stomach, thirst, difficulty swallowing, abdominal pain, nausea, vomiting, diarrhea, weak and rapid pulse, or rapid heart rate (tachycardia), shock.

**Immediate Medical Care/  
Special Treatment:**

Immediate medical attention is required. Call a physician or poison control center immediately. Treat symptomatically.

## 5. FIREFIGHTING MEASURES

**Suitable Extinguishing Media:** Dry powder, alcohol resistant foam, carbon dioxide.

**Unsuitable Extinguishing Media:** Do not use water.

**Hazardous Combustion  
Products:** Hydrogen, sulfur oxides.

**Specific Hazards:** Contact with metals may produce hydrogen gas. Excessive thermal conditions may cause decomposition, yielding sulfur oxides. Contact with water may cause violent exothermic reaction.

**Special Protective Equipment/  
Precautions for Firefighters:** As in any fire, wear MSHA/NIOSH-approved (or equivalent), self-contained, positive-pressure or pressure-demand breathing apparatus and full protective gear.

## 6. ACCIDENTAL RELEASE MEASURES

**Personal Precautions and  
Protective Equipment:** Ventilate area of leak or spill. Isolate hazard area and keep unnecessary and unprotected personnel away from the area of the leak or spill. Keep upwind. Wear appropriate personal protective equipment (see Section 8). Avoid contact with eyes, skin, and clothing.

**Emergency Procedures:** In case of chemical emergency, or if unsure how to address an accidental release, consult a professional (see Section 1).

**Methods for Containment:** Stop the flow of material, if this is without risk. Prevent entry into waterways, sewer, basements, or confined areas. Dike the spilled material, where this is possible. Product should not be released to the environment. Contain and recover liquid when possible.

**Methods for Cleanup:** Absorb spill with an inert material (e.g. vermiculite, dry sand, earth, cloth, or fleece) and place in a non-combustible container for reclamation or disposal. Do not flush to sewer. Clean contaminated surface thoroughly. Residues from spills can be diluted with water and neutralized with alkaline material such as soda ash or lime. Never return spills in original containers for reuse. Clean up in accordance with all applicable regulations.

## 7. HANDLING AND STORAGE

**Handling:** Wear personal protective equipment (see Section 8). Use only in well-ventilated areas. Provide sufficient air exchange and/or exhaust in work rooms. Avoid contact with skin, eyes, and clothing. Do not breathe vapors or spray mist. Do not ingest. When using, do not eat, drink, or smoke. Keep away from incompatible materials (see Section 10). Handle in accordance with good industrial hygiene and safety practice. Wash thoroughly after handling. Containers of this material may be hazardous when empty, as they retain product residues (vapors, liquids). Observe all warnings and precautions listed for this product. As with all acids, never add water directly to this product. Instead, add acids to water to prevent violent eruption of the solution.

**Storage:** Store in a cool, dry, ventilated area. Store in a segregated and approved area away from heat and incompatible materials (see Section 10). Store in original container. Do not store in metallic containers. Keep containers tightly closed and upright. Keep away from food, drink,

and animal foodstuffs. Keep out of the reach of children. Comply with all national, state, and local codes pertaining to the storage, handling, dispensing, and disposal of this product.

## 8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

**Exposure Limits:** Sulfuric Acid: OSHA (PEL): 1 mg/m<sup>3</sup>  
ACGIH (TLV): 0.2 mg/m<sup>3</sup>

Water: No information found.

**Engineering Controls:** Ensure adequate ventilation. Ventilation rates should be matched to conditions. If applicable, use process enclosures, local exhaust ventilation, or other engineering controls to maintain airborne levels below recommended exposure limits. If exposure limits have not been established, maintain airborne levels to an acceptable level.

### Personal Protective Measures:

**Eye/Face Protection:** Wear safety glasses with side shields or goggles and a face shield. Maintain approved eye wash station and accessible rinse facilities in work area.

**Skin Protection:** Wear appropriate chemical resistant clothing (with long sleeves) and appropriate chemical resistant gloves.

**Respiratory Protection:** An air-purifying, NIOSH-approved respirator with appropriate cartridge or canister may be permissible under certain circumstances where airborne concentrations are expected to exceed exposure limits. Use a full-face, positive-pressure, air-supplied respirator if there is any potential for an uncontrolled release, if exposure levels are unknown, or if any other circumstances exist where air-purifying respirators may not provide adequate protection.

**Specific Requirements for Personal Protective Equipment:** Ensure that glove material is compatible with this product. This information is available from glove manufacturers.

## 9. PHYSICAL AND CHEMICAL PROPERTIES

*Unless otherwise indicated, all properties are given at 25 °C and standard pressure.*

**Appearance:** Colorless, viscous, transparent liquid.

**Odor:** Very slight.

**Odor Threshold:** > 1 mg/m<sup>3</sup>

**Formula Weight:** 98.08 (as H<sub>2</sub>SO<sub>4</sub>)

**pH:** 0.3 (1 N solution)

**Melting/Freezing Point:** ≤ 3 °C

**Boiling Point/Range:** ≤ 290 °C

**Decomposition Temperature:** 340 °C

**Flash Point:** Not applicable.

**Auto-ignition Temperature:** Not applicable.

**Flammability:** Not flammable.

**Flammability/Explosive Limits:** Not applicable.

**Solubility:** Miscible with water.

**Vapor Pressure:** No information found.

<b>Vapor Density (Relative):</b>	3.4 (Air = 1)
<b>Specific Gravity:</b>	1.03 – 1.17 (Water = 1)
<b>Evaporation Rate:</b>	No information found.
<b>Viscosity:</b>	≤ 23 mPa s at 20 °C
<b>Partition Coefficient (n-octanol/water):</b>	No information found.

## 10. STABILITY AND REACTIVITY

<b>Reactivity Data:</b>	Corrosive. See Section 11.
<b>Chemical Stability:</b>	Stable under normal conditions. Sensitive to moisture.
<b>Conditions to Avoid:</b>	Excessive heat, moisture, incompatible materials.
<b>Incompatible Materials:</b>	Water, strong bases, strong acids, organic compounds, combustible materials, metals.
<b>Hazardous Decomposition Products:</b>	Hydrogen, sulfur oxides.
<b>Possibility of Hazardous Reactions:</b>	May react vigorously, violently, or explosively with the incompatible materials listed above. Excess thermal conditions may yield hazardous sulfur oxides. Contact with metals may produce hazardous concentrations of hydrogen gas. Contact with water or strong bases may cause violent exothermic reaction.
<b>Hazardous Polymerization:</b>	Will not occur.

## 11. TOXICOLOGICAL INFORMATION

<b>Routes of Exposure:</b>	Inhalation, ingestion, skin contact, eye contact.												
<b>Acute Effects:</b>	Harmful if swallowed, inhaled, or absorbed through the skin. Causes irritation of the eyes, skin, respiratory tract, and gastrointestinal tract. May enter lungs if swallowed or vomited. Liquid and vapors are corrosive. May cause tissue damage.												
<b>Chronic Effects:</b>	Prolonged or repeated exposure may affect liver function, respiratory function, kidney function, and behavioral/central nervous system function. Prolonged or repeated exposure may cause tooth decay, dermatitis, conjunctivitis, reproductive effects, mutagenic effects, and cancer.												
<b>Toxicological Data:</b>	<table> <tr> <td>Sulfuric Acid:</td> <td>LD<sub>50</sub> Oral, Rat:</td> <td>2140 mg/kg</td> </tr> <tr> <td></td> <td>LC<sub>50</sub> Inhalation, Rat:</td> <td>0.510 mg/L 2 h</td> </tr> <tr> <td></td> <td colspan="2">Corrosive to skin and eyes based on animal data.</td> </tr> <tr> <td></td> <td>Water:</td> <td>No information found.</td> </tr> </table>	Sulfuric Acid:	LD <sub>50</sub> Oral, Rat:	2140 mg/kg		LC <sub>50</sub> Inhalation, Rat:	0.510 mg/L 2 h		Corrosive to skin and eyes based on animal data.			Water:	No information found.
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	Water:	No information found.											
<b>Symptoms of Exposure:</b>	Irritation, burning, ulceration, coughing, sneezing, choking sensation, hoarseness, dyspnea, bronchitis, gastric infection, nausea, vomiting, diarrhea, thirst, difficulty swallowing, salivation, chills, fever, shock, weak and rapid pulse.												
<b>Carcinogenic Effects:</b>	This product may cause cancer.												
<b>ACGIH:</b>	Sulfuric Acid: A2 – Suspected human carcinogen												

IARC: Sulfuric Acid: 1 – Carcinogenic to humans

## 12. ECOLOGICAL INFORMATION

**Ecotoxicological Data:** Sulfuric Acid: LC<sub>50</sub>, Western Mosquitofish (*Gambusia affinis*): 42 mg/L 96 h  
EC<sub>50</sub>, Water Flea (*Daphnia magna*): 29 mg/L 24 h

Water: No information found.

**Persistence and Degradability:** Expected to be readily biodegradable.

**Environmental Effects:** Toxic to aquatic life. May leach into groundwater.

## 13. DISPOSAL INFORMATION

**Disposal Instructions:** Dispose of this material and its container to hazardous or special waste collection point. Minimize exposure to product waste (see Section 8). Do not dispose unused waste down drains or into sewers. All wastes must be handled in accordance with local, state, and federal regulations.

**Contaminated Packaging:** Because emptied containers retain product residue, follow label warnings even after container is emptied. Offer rinsed packaging material to local recycling facilities.

**Waste Codes:** D002: Waste Corrosive material (pH ≤ 2 or pH ≥ 12.5 or corrosive to steel)

## 14. TRANSPORT INFORMATION

DOT:

**UN Number:** UN2796

**Proper Shipping Name:** Sulfuric acid (with not more than 51% acid)

**Hazard Class:** 8

**Packing Group:** II

**ERG Number:** 157

**Other Transport Precautions:** No information found.

## 15. REGULATORY INFORMATION

**U.S. Federal Regulations:**

**OSHA:** This product is considered a "Hazardous Chemical" as defined by the OSHA Hazard Communication Standard, 29 CFR 1910.1200.

**TSCA Inventory:** All components of this product are on the U.S. TSCA Inventory.

**U.S. EPCRA (SARA Title III):**

**Section 302:** Sulfuric Acid: Reportable Quantity: 1000 lb  
Threshold Planning Quantity: 1000 lb

**Sections 311/312:**

Hazard Category	List (Yes/No)
Section 311 – Hazardous Chemical	Yes
Immediate Hazard	Yes
Delayed Hazard	Yes
Fire Hazard	No
Pressure Hazard	No
Reactivity Hazard	Yes

**Section 313:**

Component: Sulfuric Acid  
De Minimis Concentration: 1.0 %

**CERCLA Reportable Quantities:** Sulfuric Acid: 1000 lb

**International Inventories:**

Country or Region	Inventory Name	On Inventory (Yes/No)*
Australia	Australian Inventory of Chemical Substances (AICS)	Yes
Canada	Domestic Substances List (DSL)	Yes
Canada	Non-Domestic Substances List (NDSL)	No
China	Inventory of Existing Chemical Substances in China (IECSC)	Yes
Europe	European Inventory of Existing Commercial Chemical Substances (EINECS)	Yes
Europe	European List of Notified Chemical Substances (ELINCS)	No
Japan	Inventory of Existing and New Chemical Substances (ENCS)	Yes
Korea	Existing Chemicals List (ECL)	Yes
Philippines	Philippine Inventory of Chemicals and Chemical Substances (PICCS)	Yes

\*A "Yes" indicates that the listed components of this product comply with the inventory requirements administered by the governing country or region.

<b>16. OTHER INFORMATION</b>
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**Issue Date:**

July 10, 2023

**Reason for Revision:**

Updated product identifiers in Section 1. Supersedes 12/22/2021 version.